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chain alkanes.

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Examples, Ch. 21: 11-14, 17-19, 21, 24, 27-30, 59-60, 73 II. Rates of Decay (21.4): Know the half-life equation for radioactive decay (first order reaction) and be able to use it to convert between amount and time, or vice versa. Examples, Ch. 21: 34-42, 80 III. Energy Changes (21.6): Know how to calculate the energy change for a nuclear reaction

## **Study Guide for Exam #3, Ch. 19 21**

Chapter 21 Nuclear Chemistry Study Guide 1 Topic . Expand. Lesson Content . 0% Complete 0/1 Steps. Ch20 Nuclear Chemistry Chapter Test (15 Questions) 21 - NUCLEAR CHEMISTRY 9 Topics . Expand. Lesson Content . 0% Complete 0/9 Steps. 21.1 Introduction to Nuclear Chemistry and Trends in Radioactivity (10:05)

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Chemistry Honors Ch ( 1-7.1 )-Midterm study Guide Choose the best answer from the options that follow each question. \_\_\_A 1. A scientist is working in the lab, with compounds containing carbon. She is most likely working in the field of The answer is A because organic chemistry is the study of most carbon-containing compounds. \_A\_\_\_ 2.

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